## DEPARTMENT OF CHEMISTRY UNIVERSITY OF PENNSYLVANIA MCE PROGRAM CHEMISTRY 503, 2006/7

## WORKSHOP PROBLEMS March 31st

## **ANSWERS**

## Dr. Roberts

3. If thymine existed predominantly in the tautomeric form in which the carbonyl group at C4 is enolized (see Table 26.1, p741, in MC for the numbering of the ring), explain with the aid of appropriate structures how this would disrupt the AT base pair.

the AT base pair

H-bonding no longer possible here

the AT base pair if the C4 carbonyl group of T is enolized